

Deep cryogenic treatment in AISI D2 tool steel punches to enhance tool life

Fabiano Dornelles Ramos ^{1,2*} 

Josué de Oliveira Bairros ³

Mauro Francisco Castro Moscoso ^{1,4}

Abstract

New materials and heat treatment are being developed worldwide to improve tool life. Tool steels are expensive due to the alloying elements and fabrication costs. Therefore, the maximization of tool life keeping good surface finishing is especially important to keep the processes attractiveness. The application of cryogenic treatment is promising to enhance mechanical and metallurgical properties of cold work tool steel parts. The possibility of increasing the finer secondary carbides content and decreasing amount of retained austenite during deep cryogenic treatment (DCT) is also an important matter for industry. The aim of this work is to improve tool life by applying DCT in AISI D2 cutting punches and compare these results with punches treated by conventional heat treatment (CHT), vacuum furnace quenching plus triple tempering. The tests used for analyses were: metallography, X-ray diffraction, carbides fraction, hardness and number of parts produced by each punch (practical test). The results indicate that the DCT punches have an enhancement of approximately 50% in its useful life.

Keywords: Deep Cryogenic; Treatment; Tool life; Tool steel.

1 Introduction

The need to increase productivity without raising production costs leads some sectors of the industry to look for ways to address these needs. Because of this, studies aimed at the development of new technologies are being developed. Also, the manipulation of existing processing methods to increase tool life and thereby increase productivity and reduce costs are being implemented [1-5].

AISI D2 tool steel has high wear and abrasion resistance. High alloying content may provide excellent hardenability and good dimensional stability [6] since the parts are suitable to air quenching. This grade of steel is widely used for cold working, especially for cutting tools and dies production [7, 8]. Since this tool steel has greater alloying elements content, some amount of retained austenite can be formed [9], depending on the heat treatment performed [10]. The microstructure of high hardness martensite matrix, with large primary carbides and very small secondary carbides promotes high wear resistance. Therefore, a cryogenic treatment is recommended by the steel producers to reduce the amount of retained austenite in high carbon steels [11-13].

Conventional quenching and tempering have been used to improve the mechanical properties of metals;

however, cryogenics is a relatively recent process that is capable of not only raising certain properties but maximizing them [8,14-16]. This cryogenic process aims to improve the mechanical properties of quenched and tempered steels, thus improving wear, mechanical resistance, and hardness through modifying the distribution of carbides and phases formed after treatment [5,17].

This work aims to apply DCT to AISI D2 cutting punches and evaluate tool life in comparison to conventional cutting punches with CHT. Quenching was performed in a vacuum furnace with nitrogen blow and the cryogenic treatment was carried out by immersion in a liquid nitrogen tank. Furthermore, the samples were evaluated by chemical analysis, hardness, metallography (qualitative and quantitative), X-ray diffraction and practical test.

2 Materials and methods

The experimental procedure followed the steps presented in Figure 1.

¹Instituto Federal de Educação, Ciência e Tecnologia do Rio Grande do Sul, Caxias do Sul, RS, Brasil.

²Programa de Pós-graduação em Engenharia de Minas, Metalúrgica e Materiais – PPG3M, Universidade Federal do Rio Grande do Sul – UFRGS, Porto Alegre, RS, Brasil.

³Fras-le S.A, Caxias do Sul, RS, Brasil.

⁴Termo Aço Tratamentos Térmicos LTDA, Caxias do Sul, Brasil.

⁵Programa de Pós-graduação em Tecnologia e Engenharia de Materiais – PPG-TEM, Instituto Federal de Educação, Ciência e Tecnologia do Rio Grande do Sul, Caxias do Sul, Brasil.

*Corresponding author: fabiano@caxias.ifrs.edu.br; fdrpunk@gmail.com



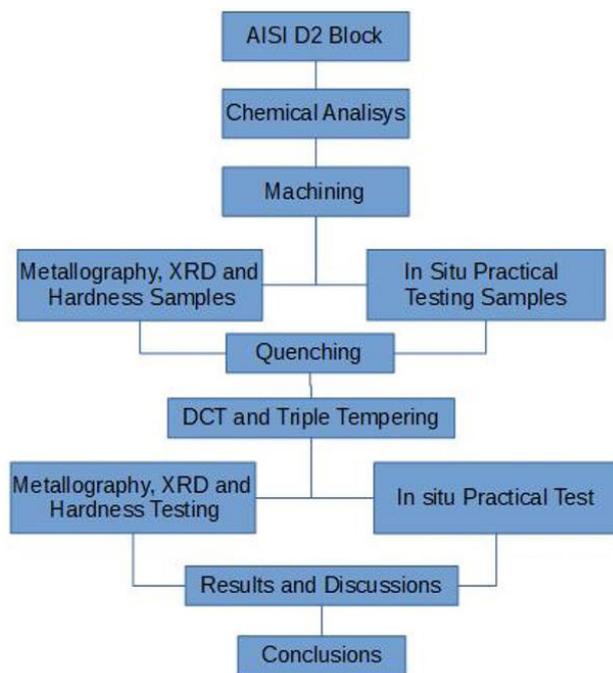


Figure 1. Block diagram showing the experimental procedure.

Samples were produced from two blocks of AISI D2 steel with dimensions 45mm thick, 300mm wide and 300mm long. Cutting punches were made for stamping of plates. The material is a cold work tool steel AISI D2, with the chemical composition according to Table 1. Chemical analysis was tested by optical emission spectrometer from Spectro, model Spectrolab (average of 3 measurements).

Table 1. Chemical analysis of the sample

Chemical Composition of AISI D2 [wt %]							
[C]	[Mn]	[P]	[S]	[Si]	[Cr]	[Mo]	[V]
1.51	0.35	0.018	0.009	0.43	12.00	0.92	0.96

The tool was produced by milling, flat grinding and wire EDM (electric discharge machining) processes. Besides that, samples were produced to metallography, hardness testing and X-ray diffraction.

2.1 Heat treatment

For the heat treatment of the samples and the tool a vacuum furnace was employed. The austenitizing temperature and time was 1030°C, 1h/inch. After quenching, samples were submitted to two types of treatment: Conventional Heat Treatment (CHT) and Deep Cryogenic Treatment (DCT).

CHT consists in triple tempering after quenching in an electric convection furnace, at 530°C for 2h each.

DCT consists in immersion in liquid nitrogen for 12h, following by single tempering in an electric convection furnace, at 530°C for 2h.

2.2 Metallographic examination, Drx and hardness testing

For metallographic examination samples were cut by abrasive cutting machine and mounted in Bakelite to better preparation. Metallography followed standard procedure [18] of grinding, polishing, and etching, followed by Optical Microscopy. Besides, to estimate the percentage of carbides the software ImageJ® was used. Images at 500X magnification were used to measuring carbides fraction area. Hardness measurements were made in a Rockwell C hardness tester (average of 3 tests). XRD measurements were performed on a flat, polished surface to evaluate the phases in the treated samples.

2.3 Practical test

To assess the behaviour of DCT and compare to CHT in the industry, two tools were produced and put into work with the respective treatments. The parameter used to compare the performance of the tools was the tool life between two sharpening. The results and discussion are presented in the next topic.

3 Results and discussion

3.1 Metallography and X-ray diffraction

Metallographic examination shows the presence of tempered martensite [matrix], primary carbides (PC) and secondary carbides (SC). According to Das et al. [2] and Hadi Ghasemi-Nanasa [19] Primary carbides consists in M_7C_3 stoichiometry. The secondary carbides could be M_2C and $M_{23}C_6$. Large primary carbides are formed during solidification of the steel, while secondary carbides precipitates from solid solution, this why they are smaller and, therefore, implies in higher mechanical properties enhancement. Also, secondary carbides can be divided into two groups, large secondary carbides (LSC) and small secondary carbides (SSC), as shown in Figure 2.

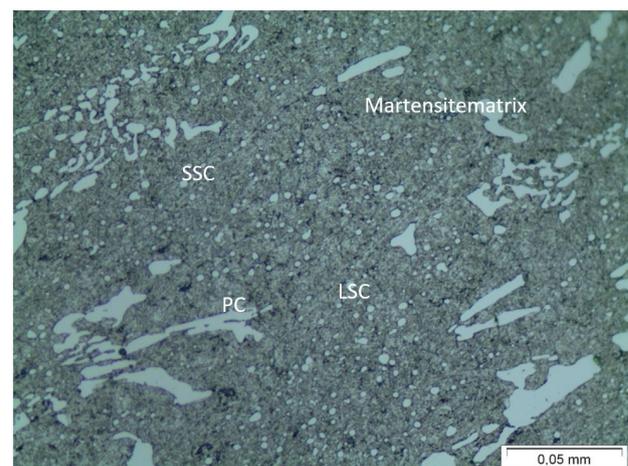


Figure 2. Micrograph of CHT sample, where the martensite matrix, PC, SSC and LSC can be observed.

Figure 3 shows the microstructure observed for the CHT and DCT heat treatments. It is possible to observe that secondary carbides quantity is higher for DCT in comparison to the CHT samples

According to literature [5,20] the higher quantity of SSC and LSC comes from the precipitation during tempering and cryogenic stages. The carbon trapped in the martensite is released and reacts with the chromium alloying element. This kind of precipitation has coherence with the matrix lattice, therefore, higher influence on mechanical properties.

To estimate the carbides fraction image analyses techniques were applied to 500X magnification micrographs. Figure 4 shows the comparison between CHT and DCT samples.

In his study, Vitry et al. [21] suggested that the carbides in AISI D2 steel may have three types, as shown in Figure 5.

In general, in the micrographs of AISI D2 steel, a tempered martensite matrix with carbides of types MC and $M_{23}C_6$ dispersed in the matrix is observed. It appears that carbides of the type MC are located in the centre or at the prior austenite grain boundaries, as show in Figure 5. The presence of M_7C_3 carbides was also observed, this kind of primary carbide can present different morphology due to

the hot rolling process. According to Vitry et al. [21], this type of carbide appears preferentially in regions subjected to intense cooling.

ImageJ was used to estimate carbides fraction, the results are shown in Table 2.

To estimate of the percentage area of carbides, only secondary carbides were counted. It is possible to observe that DCT implies in an increase about 64.8% of carbides percentage. According to Das et al. [2-4] and Moscoso et al. [22] the carbon atoms trapped into martensite lattice are expelled during DCT for the surrounding dislocations. This carbon combined with the alloying elements forms transition carbides that, over time, become finely precipitated secondary carbides of the $M_{23}C_6$ type. Also, the conditioning of the retained austenite implies in more carbon to precipitate finer secondary carbides, since in this temperature the diffusion and energy to increase the size of carbides is not enough.

Table 2. Calculation of carbides percentual area. Samples CHT and DCT

Sample	% SECONDARY CARBIDES AREA
CHT	6.6
DCT	10.1

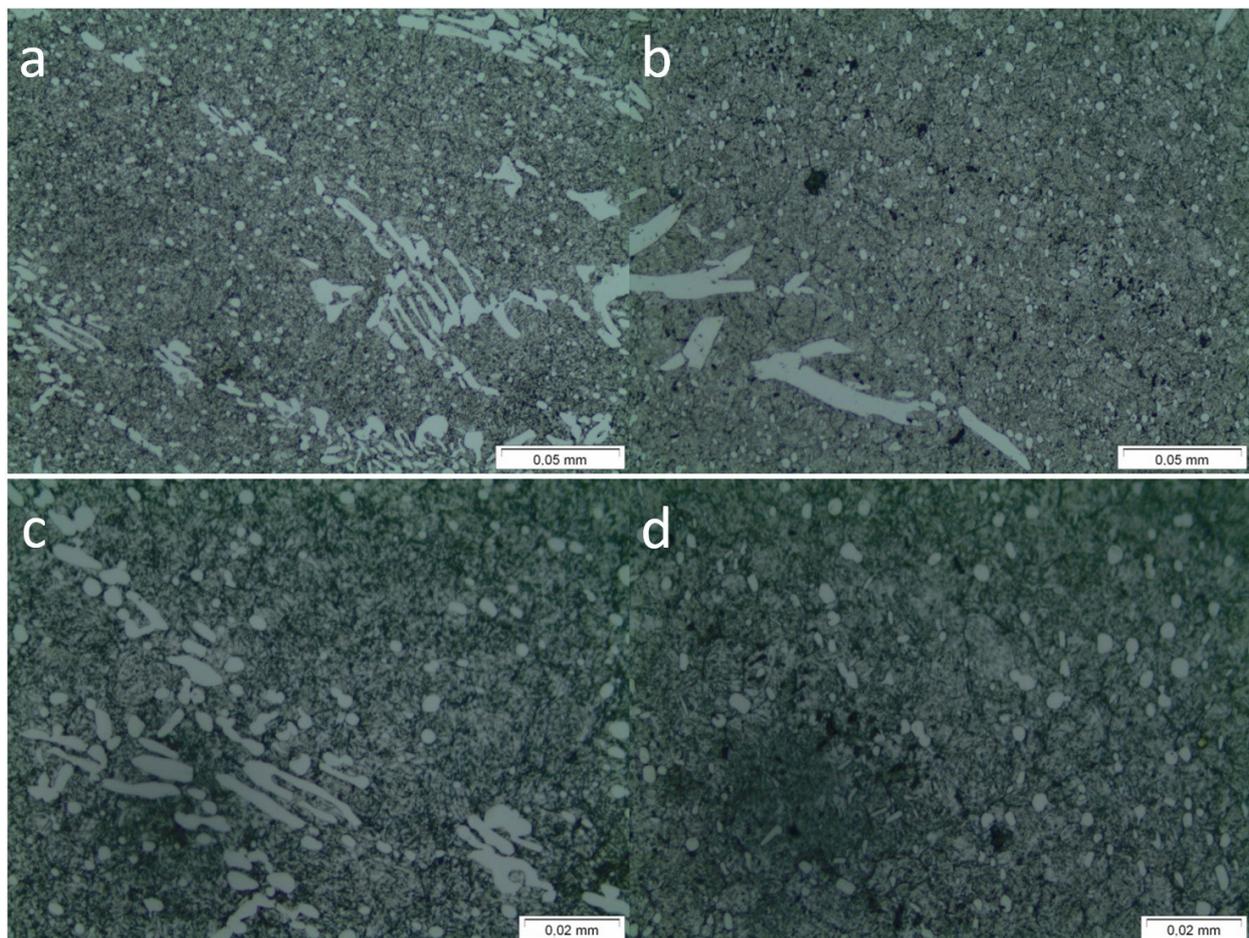


Figure 3. Lower magnification photomicrography of (a) CHT samples and (b) DCT sample. Higher magnification photomicrography of (c) CHT samples and (d) DCT sample.

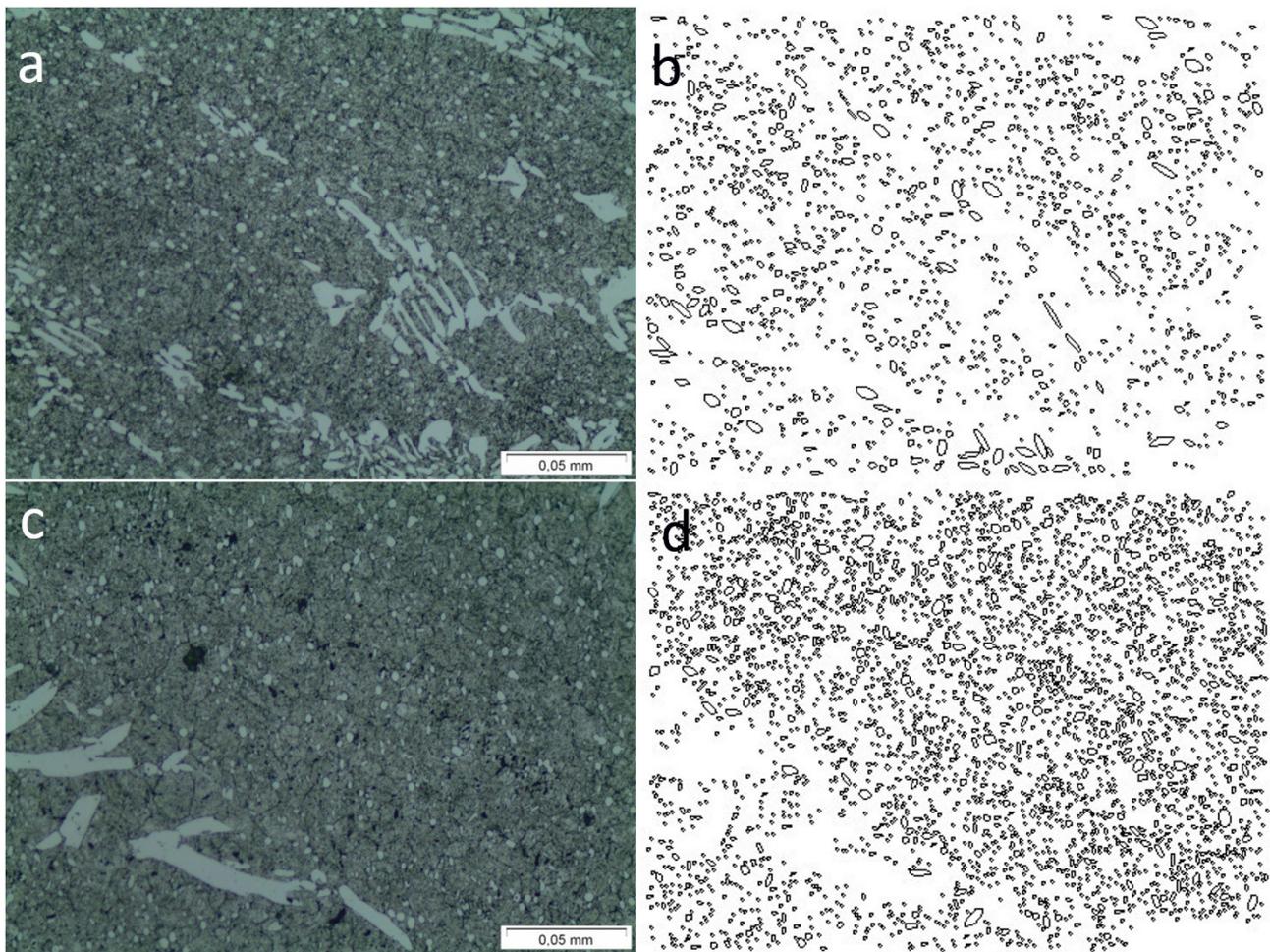


Figure 4. Binarization using ImageJ Software. (a) and (b) Sample CHT. (c) and (d) Sample DCT.

Carbide type	Morphology	Preferential Location
MC	 Globular, isolated or associated	Grain center or boundary, also associated to M7C3
M ₇ C ₃	 Thick lamellar like fish spine	Intense cooling regions
M ₂₃ C ₆	 Small isolated globular shape	Homogeneously distributed in matrix

Figure 5. Carbides morphology and preferential regions in tool steels. Adapted from Vitry et al. [21]

M₂₃C₆ carbides of small and globular morphology show a better distribution in samples submitted to cryogenic and subsequent tempering. This observation is in line with what Amini et al. [23,24] states in their article. According to the authors, cryogenic cooling facilitates the precipitation of carbides in greater proportion and better dispersed in the matrix. This is due to the contraction of the crystalline structure of metal during cryogenic treatment, promoting the diffusion of carbon atoms to dislocations or discontinuities. Figure 6 shows the results of XRD for both samples.

The intensity of Cr₂₃C₆ peaks obtained was not so high, this is probably due to the use of solid samples. Also some peaks were dislocated, probably due to residual stresses from the heat treatment. Still, it is possible to observe higher peaks of Cr₂₃C₆ in DCT samples, comparing to CHT samples. This is an indication that these carbides quantity are higher in DCT samples. Also, austenite and Cr₇C₃ peaks looks very similar, indicating that the amount of each phase is similar comparing both heat treatments.

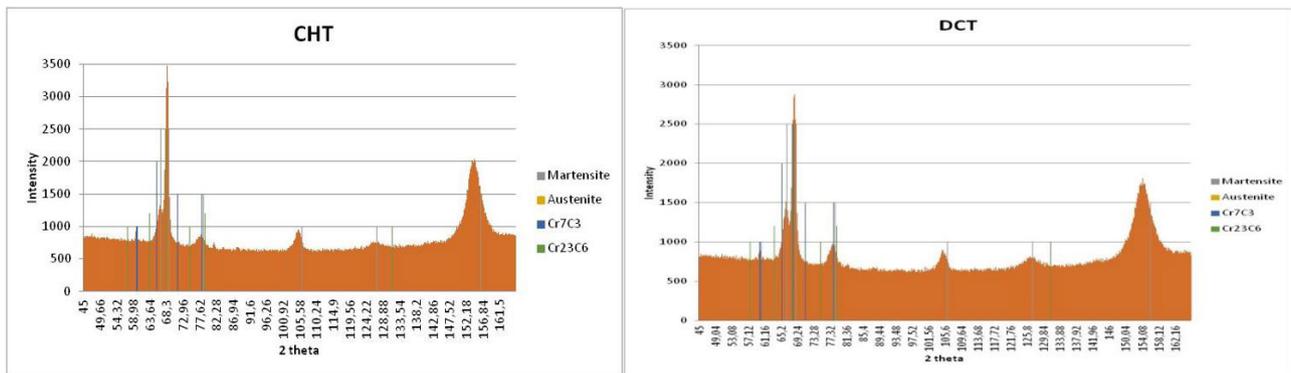


Figure 6. X-ray diffraction for CHT and DCT samples.

3.2 Hardness measurements

Hardness measurements shows a significant increment in hardness after DCT, the reason is probably by transforming retained austenite into martensite [25], that was not possible to observe by light microscopy. Further tempering, reduced hardness to values close to CHT. It is important to observe that the presence of retained austenite not only decreases the hardness but, also implies in the possibility of transformation in martensite during work. This non tempered martensite promotes distortion and also reduces tool life due to brittleness, as shown in Figure 7.

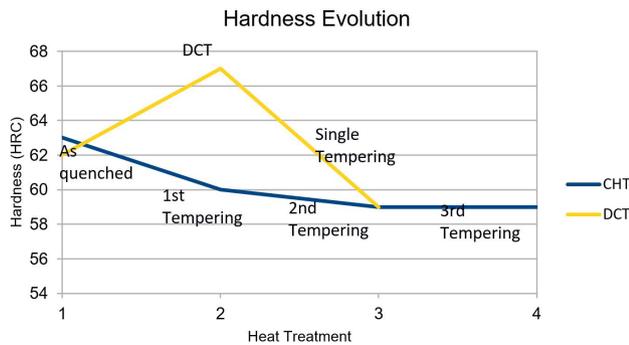


Figure 7. Hardness evolution after CHT and DCT.

3.3 In Situ practical testing

To develop the practical test, cutting punches were treated by CHT and DCT and placed to stamp platelets at a large automotive company. The stopping criterion used was the formation of burrs on the produced part. Treated cutting punches were used at random days. The industrial application showed that the DCT samples showed superior tool life in all tests. The average number of pieces made for the DCT samples was 8050.5, while for the CHT samples it was 5353.7. That means an increment of 50.3% in tool life. The formation of a high resistance martensitic matrix with the formation of fine, well distributed, carbides results in higher toughness [26]. The results of industrial application are shown in Figure 8.

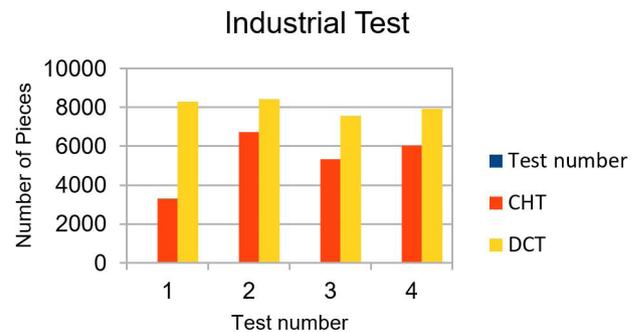


Figure 8. Industrial application of cutting punches.

4 Conclusions

From the results obtained by this work it is possible to conclude that:

1. It was possible to improve tool life by applying deep cryogenic treatment to cutting punches of AISI D2 tool steel;
2. The use of DCT produced higher carbide precipitation when compared to CHT;
3. Tool life improvement was about 50% comparing to conventional heat treatment;
4. It was possible to improve steel properties with a single tempering after DCT. It implies in energy cost reduction.

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